

## Chemical Evolution of the Universe

### Problem sheet 3

1. (a) Consider a gas cloud consisting only of  $^1\text{H}$ ,  $^4\text{He}$  and  $^7\text{Li}$ . Given the abundance of  $^7\text{Li}$  relative to  $^1\text{H}$  by number,  $\frac{n_{\text{Li}}}{n_{\text{H}}} = 10^{-6}$ , and the abundance of  $^4\text{He}$  by mass,  $Y = \frac{\text{mass of He}}{\text{total mass}} = 0.25$ , what is the abundance of  $^4\text{He}$  relative to  $^1\text{H}$  by number,  $y = \frac{n_{\text{He}}}{n_{\text{H}}}$ ?

(b) Show that  $Y$  and  $y$  are related by

$$Y = \frac{4y(1-Z)}{1+4y},$$

where  $Z = \frac{\text{mass of metals}}{\text{total mass}}$  is the abundance by mass of all metals.

**3 points**

2.  $^4\text{He}$  is used in a range of medical, industrial and scientific applications. So its price actually matters to our everyday lives. Estimate the price (in US\$) of a cubic metre of  $^4\text{He}$  if the neutron lifetime were 100 s instead of 882 s.

Hint 1: Assume a linear relationship between the price of  $^4\text{He}$  and its availability (by mass).

Hint 2: <https://minerals.usgs.gov/minerals/pubs/commodity/helium/mcs-2017-heliu.pdf>

**2 points**

3. In problem 3. of problem sheet 2 we learned that the small baryon-to-photon ration,  $\eta$ , is the reason that the Universe has to cool to a temperature significantly below that corresponding to the binding energy of deuterium in order for deuterium formation to set in. We will now derive the same result in a different way.

- (a) First, recall from the lecture the formula for the number density of a non-relativistic particle species. Write down the number densities of neutrons and protons,  $n_n$  and  $n_p$ , remembering that in some places you can use  $m_n \approx m_p \approx m_b$  (but not in others).
- (b) Now consider a nucleus X of mass number  $A$  and charge  $Z$ . Write down its number density  $n_X$ , remembering that in some places you can use  $m_X \approx Am_b$  (but not in others).
- (c) The chemical potential is conserved in the reactions that produces nucleus X from  $Z$  protons and  $(A - Z)$  neutrons, such that  $\mu_X = Z\mu_p + (A - Z)\mu_n$ . Use this fact to eliminate  $\mu_X$  from your result in (b).
- (d) Finally, show that

$$\frac{n_X}{n_b} = f(A) \left( \frac{kT}{m_b c^2} \right)^{3/2(A-1)} \eta^{A-1} \left( \frac{n_p}{n_b} \right)^Z \left( \frac{n_n}{n_b} \right)^{A-Z} \exp\left( \frac{\text{BE}_X}{kT} \right)$$

where  $f(A)$  is a dimensionless number of order a few that depends only on  $A$  and  $g_X$ , and  $\text{BE}_X$  is the binding energy of nucleus X. Remember that  $n_b = \eta n_\gamma$ , and recall the result of problem 3.(b) on problem sheet 2.

Can you now see why  $kT$  has to drop significantly below  $\text{BE}_X$  before the abundance of X can rise to appreciable levels?

**8 points**